

MINOR POLYHYDROXYLATED STEROLS FROM THE STARFISH  
*PROTOREASTER NODOSUS*<sup>1</sup>

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**ABSTRACT.**—Three minor polyhydroxylated sterols—5 $\alpha$ -cholestane-3 $\beta$ ,4 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-heptol (**4**), 24 $\xi$ -methyl-5 $\alpha$ -cholestane-3 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-hexol (**5**), and (22E)-24 $\xi$ -methyl-5 $\alpha$ -cholest-22-en-3 $\beta$ ,4 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-heptol (**6**)—have been isolated from the Pacific starfish *Protoreaster nodosus*.

The number of known polyhydroxylated steroids from marine animals is steadily growing. They have been isolated from soft corals, gorgonians, nudibranchs, and starfish (1-6). Recently, we have described the occurrence of three highly hydroxylated steroids, with moderate cytotoxicity, from the starfish *Protoreaster nodosus* L. (7).

A common feature of these compounds is the 3 $\beta$ ,6 $\alpha$ ,8 $\beta$ ,15 $\alpha$ ,16 $\beta$ ,26-hexahydroxy moiety, one of them is simply the hexol (**1**), the second is related to **1** with the introduction of an additional hydroxyl group at 7 $\alpha$ -position (**2**), and the third is the octol (**3**) with two additional hydroxyl groups at the 7 $\alpha$ - and 4 $\beta$ -positions.

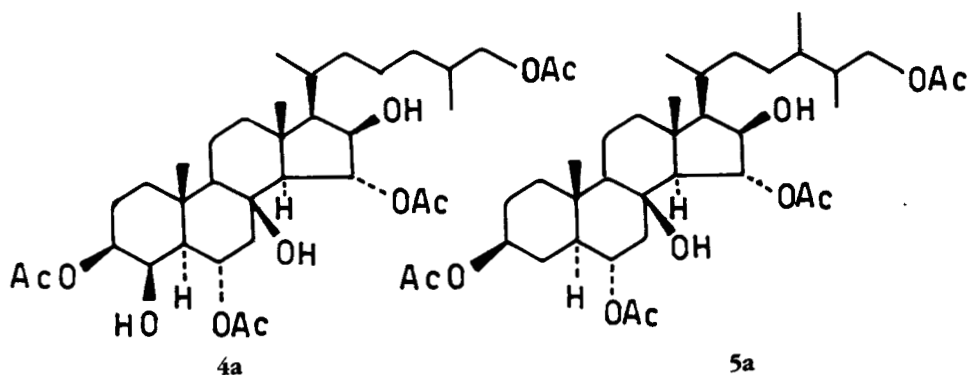
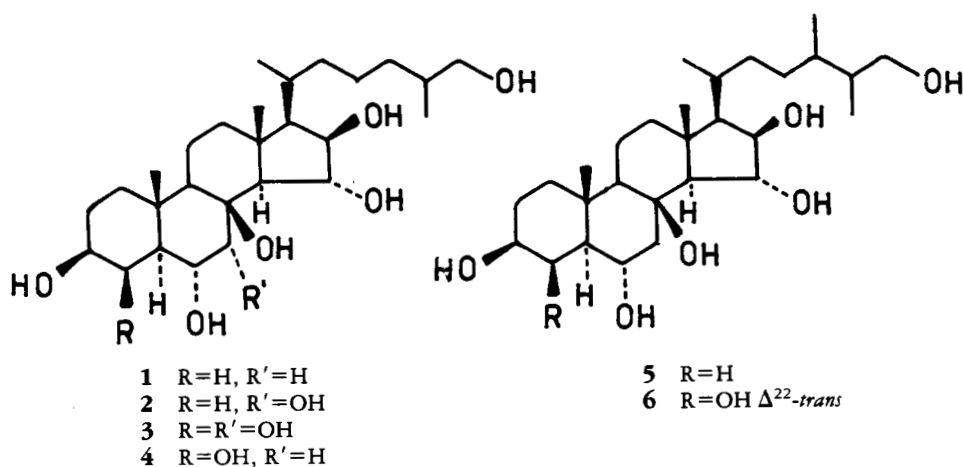
We now wish to report the isolation of three very minor hydroxylated sterols, 5 $\alpha$ -cholestane-3 $\beta$ ,4 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-heptol (**4**), 24 $\xi$ -methyl-5 $\alpha$ -cholestane-3 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-hexol (**5**), and (22E)-24 $\xi$ -methyl-5 $\alpha$ -cholest-22-en-3 $\beta$ ,4 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-heptol (**6**) from *P. nodosus*.

## EXPERIMENTAL

The animals (*P. nodosus*, identified by P. Laboute and P. Tirard, ORSTOM de Nouméa) were collected off Nouméa, Nouvelle Calédonie in August 1981, and lyophilized (2 kg). The lyophilized animals were extracted in a Soxhlet apparatus with light petroleum ether (bp 40-70°), then with MeOH-CHCl<sub>3</sub> (1:9), followed by MeOH. The MeOH-CHCl<sub>3</sub> extract (50 g) was extracted again with EtOAc and then MeOH. The MeOH extract (22 g) was separated by short column chromatography using 500 g of silica gel (15 $\mu$ ) in CHCl<sub>3</sub>-MeOH (9:1) and increasing amounts of MeOH. The fractions eluted with CHCl<sub>3</sub>-MeOH (3:1), and (7:3) gave 1.6 g and 1.1 g of residue, respectively, which were combined and chromatographed on Sephadex LH-60 (90 $\times$ 3 cm, 58 g of resin) by using MeOH as eluent. Fractions of 10 ml were collected and monitored by SiO<sub>2</sub>-tlc in CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (80:18:2). The fractions 40-58 (1.1 g) were rechromatographed on Sephadex LH-20 and 7.5 ml fractions were eluted with MeOH. Fractions 45-61 gave 0.49 g of residue, which was submitted to droplet counter current chromatography (DCC-A apparatus, Tokyo Rikakikai, 300 tubes, solvent system: CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (7:13:8), flow rate 12 ml/h, ascending mode; the eluants were collected in 5 ml fractions and monitored by tlc-SiO<sub>2</sub>). Fractions 65-78 (34 mg) contained mainly **4** and the glycoside nodososide (8); fractions 90-93 (16 mg) contained mainly **5** and small amounts of **6** and the previously reported **1** (7); fractions 111-124 contained **6** and major amounts of the previously reported **2** (7). Each of these DCC fractions was chromatographed by hplc on a Whatman Partisil M-9 10/50 ODS column (eluant, MeOH-H<sub>2</sub>O, 7:3; M6000 pump, R401 refractometer, U6K injector, all from Waters) to give the individual minor hydroxylated sterols, **4** (9 mg), **5** (8 mg), and **6** (1.5 mg), with elution times of 13.2, 17.2, and 14.2 min, respectively.

**Compound 4:** mp 241-243° (from MeOH in the presence of CH<sub>2</sub>Cl<sub>2</sub> vapors); [ $\alpha$ ]<sub>D</sub><sup>20</sup> = +28.4 (c, 1, MeOH); eims (70 eV) *m/z* (%) 466 (M<sup>+</sup>-H<sub>2</sub>O, 8), 448 (62), 430 (37), 412 (15), 347 (10), 337 (M<sup>+</sup>-side chain, 15), 319 (32), 301 (30), 283 (42), 265 (25), 241 (*a*, 42), 225 (*b*-H<sub>2</sub>O, 100), 223 (25) and 207 (60); pmr (CD<sub>3</sub>OD)  $\delta$  0.94 (d, *J* = 6.5 Hz, CH<sub>3</sub>-21), 0.96 (d, *J* = 6.5 Hz, CH<sub>3</sub>-27), 1.14 (s, CH<sub>3</sub>-18), 1.22 (s,

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CH<sub>3</sub>-19), 2.50 (dd,  $J=13$  and 3Hz, 7 $\beta$ -H), 3.43-3.49 (2H, one dd with  $J=10.5$  and 6.0 Hz superimposed on a broad signal, 26- and 3 $\alpha$ -H), 4.01 (dd,  $J=7.5$  and 3Hz, 16 $\alpha$ -H), 4.09 (2H, one dd with  $J=10.5$  and 3Hz, 15 $\beta$ -H superimposed on the signal for 6 $\beta$ -H), 4.28 (br s,  $W_{1/2}=6$ Hz, 4 $\alpha$ -H).

**Compound 5**: mp 253-256° (from MeOH in the presence of CH<sub>2</sub>Cl<sub>2</sub> vapors);  $[\alpha]_D^{25} = +27.8^\circ$  (c, 0.8, MeOH); eims (70 eV)  $m/z$  (%) 464 (M<sup>+</sup> -H<sub>2</sub>O, 62), 449 (8), 446 (44), 431 (20), 428 (30), 413 (22), 410 (11), 331 (20), 321 (M<sup>+</sup> -side-chain, 23), 303 (40), 285 (38), 267 (26), 239 (b -H<sub>2</sub>O, 100), 225 (a, 45), 221 (b-2H<sub>2</sub>O, 60); pmr (CD<sub>3</sub>OD)  $\delta$  0.82 (d,  $J=6.5$ Hz, CH<sub>3</sub>-26 or CH<sub>3</sub>-27), 0.84 (d,  $J=6.5$ Hz, CH<sub>3</sub>-27 or CH<sub>3</sub>-28), 0.95 (d,  $J=6.5$ Hz, CH<sub>3</sub>-21), 1.05 (s, CH<sub>3</sub>-19), 1.15 (s, CH<sub>3</sub>-18), 2.43 (dd,  $J=12$  and 3Hz, 7 $\beta$ -H), 3.40 (1H, dd,  $J=10.5$  and 6Hz, 26-H), 3.51 (broad m, 3 $\alpha$ -H), 3.64 (dt,  $J=3$  and 10Hz, 6 $\beta$ -H), 4.01 (dd,  $J=8.0$  and 2.5 Hz, 16 $\alpha$ -H), 4.06 (dd,  $J=11.0$  and 2.5Hz, 15 $\beta$ -H).

**Compound 6**: mp 185-188° (from MeOH in the presence of CH<sub>2</sub>Cl<sub>2</sub> vapors),  $[\alpha]_D^{25} = +20.0^\circ$  (c, 0.4, MeOH); eims (70 eV)  $m/z$  (%) 478 (M<sup>+</sup> -H<sub>2</sub>O, 25), 460 (100), 442 (60); pmr (CD<sub>3</sub>OD)  $\delta$  0.91 (d,  $J=6.5$ Hz, CH<sub>3</sub>-27), 0.99 (d,  $J=6.5$ Hz, CH<sub>3</sub>-28), 1.045 (d,  $J=6$ Hz, CH<sub>3</sub>-21), 1.17 (s, CH<sub>3</sub>-18), 1.22 (s, CH<sub>3</sub>-19), 2.10 (m, H-24), 2.50 (dd,  $J=13.5$  and 3.5Hz, 7 $\beta$ -H), 2.55 (m, 20-H), 3.41 (2H, one dd with  $J=10.5$  and 6Hz, partially superimposed on a broad signal, 26-H and 3 $\alpha$ -H), 3.56 (1H, dd,  $J=10.5$  and 6.5Hz, 26-H), 3.93 (dd,  $J=7.5$  and 2.5Hz, 16 $\alpha$ -H), 4.09 (2H, one dd with  $J=11.0$  and 2.5Hz, 15 $\beta$ -H superimposed on the 6 $\beta$ -H signal), 4.28 (m,  $W_{1/2}=6$ Hz, 4 $\alpha$ -H), 5.45 (2H, m, 22 and 23-H).

**5 $\alpha$ -Cholestane-3 $\beta$ ,4 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-heptol 3,6,15,26-tetraacetate 4a**: The mixture of **4** (3 mg) and excess of Ac<sub>2</sub>O in 0.1 ml of dry pyridine was kept at room temperature overnight. The excess reagents were evaporated under vacuum, and the residue pyridine was removed by coevaporation with C<sub>6</sub>H<sub>6</sub>. The dark brown polar materials were removed by dissolving the residue in CHCl<sub>3</sub> and passing it through a Pasteur pipette fitted with a slurry of a silica gel in CHCl<sub>3</sub>. The eluate was evaporated to give a residue (2.3 mg) containing the tetraacetate, **4a**. Eims (70 eV) ( $m/z$ ) (%) 592 (M<sup>+</sup> -CH<sub>3</sub>CO<sub>2</sub>H, 15), 574 (10), 532 (18), 514 (55), 496 (18), 472 (25), 454 (M<sup>+</sup> -3CH<sub>3</sub>CO<sub>2</sub>H-H<sub>2</sub>O, 100); pmr (CDCl<sub>3</sub>)  $\delta$  0.91 (d,  $J=6.5$ Hz, CH<sub>3</sub>-21 or CH<sub>3</sub>-27), 0.92 (d,  $J=6.5$ Hz, CH<sub>3</sub>-27 or CH<sub>3</sub>-21), 1.19 (s, CH<sub>3</sub>-18), 1.29 (s, CH<sub>3</sub>-19), 2.05, 2.09, 2.10 and 2.11 (four s, 12H, O=C-CH<sub>3</sub>), 3.83 (1H, dd,  $J=10.0$  and 6.5Hz, 26-H), 3.96 (3H, br signal, 4 $\alpha$ -, 16 $\alpha$ - and 26-H), 4.72 (2H, broad signal, 3 $\alpha$ - and 15 $\beta$ -H), 5.28 (1H, dt,  $J=4$  and 10.5Hz, 6 $\beta$ -H).

24 $\xi$ -Methyl-5 $\alpha$ -Cholestane-3 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-hexol 3,6,15,26-tetraacetate **5a**: This compound was prepared and purified by the method described above; eims (70 eV) ( $m/z$ ) (%) 590 ( $M^+$ -CH<sub>3</sub>CO<sub>2</sub>H, 35), 582 (10), 530 (60), 512 ( $M^+$ -2CH<sub>3</sub>CO<sub>2</sub>H-H<sub>2</sub>O, 100), 494 (20), 470 (45), 452 (70); pmr (CDCl<sub>3</sub>)  $\delta$  0.80 (d,  $J=6.5$ Hz, CH<sub>3</sub>-28 or CH<sub>3</sub>-27), 0.84 (d,  $J=6.5$ Hz, CH<sub>3</sub>-27 or CH<sub>3</sub>-28), 0.90 (d,  $J=7.0$ Hz, CH<sub>3</sub>-21), 1.09 (s, CH<sub>3</sub>-19), 1.19 (s, CH<sub>3</sub>-18), 2.03, 2.04, 2.05, 2.07 (four s, 12H, O=C-CH<sub>3</sub>), 3.88 (1H, dd,  $J=10.5$  and 7.0Hz, 26-H), 3.99 (2H, broad m, 26 and 16 $\alpha$ -H), 4.69 (2H, one dd,  $J=12$  and 2.5Hz emerging from a broad m, 15 $\beta$  and 3 $\alpha$ -H), 4.92 (dt,  $J=4$  and 12Hz, 6 $\beta$ -H).

HYDROGENATION OF **6**.—The  $\Delta^{22}$ -unsaturated sterol **6** (1 mg) was hydrogenated for 12 h at room temperature and atmospheric pressure using Pd/C (5%, 1 mg) as catalyst and MeOH as solvent. Filtration and evaporation of the solvent gave a residue whose pmr showed in the methyl region the following signals,  $\delta$  0.82 (d,  $J=6.5$ Hz), 0.85 (d,  $J=6.5$ Hz), 0.95 (d,  $J=7.0$ Hz), 1.14 (s, CH<sub>3</sub>-18), 1.22 (s, CH<sub>3</sub>-19); the olefinic signal had disappeared.

INSTRUMENTAL.—Pmr and cmr spectra were recorded on Bruker WX-270 and WM-250 instruments. Eims were obtained on an AEI MS-30 mass spectrometer. Optical rotation was measured on a Perkin Elmer 141 polarimeter in MeOH. Mps were measured on a Kofler hot-stage apparatus and are uncorrected.

## RESULTS AND DISCUSSION

5 $\alpha$ -Cholestane-3 $\beta$ ,4 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-heptol (**4**), had mp 241-243°, [ $\alpha$ ]<sub>D</sub> = +28.4°. In the eims the highest molecular weight ion observed ( $m/z$  466) corresponded to loss of H<sub>2</sub>O from the molecular formula C<sub>27</sub>H<sub>48</sub>O<sub>7</sub>. The fragmentation pattern, with ions for stepwise H<sub>2</sub>O loss ( $m/z$  448, 430, and 412) and ions corresponding to the loss of an hydroxylated C<sub>8</sub> side chain ( $m/z$  337) together with H<sub>2</sub>O loss ( $m/z$  319, 301, 283, and 265), closely resembled that observed in the spectrum of the heptol **2** (7). The spectrum also contained two intense ions at  $m/z$  241 (42%) and 225 (100%), each accompanied by loss of H<sub>2</sub>O ( $m/z$  223 and 207). An intense peak at  $m/z$  225 was already observed in the hexol **1** and in the heptol **2**; in the spectrum of this latter a peak at  $m/z$  223 (i.e., 241-H<sub>2</sub>O) was also present. We propose that these fragments can originate by cleavages of the 12, 13 and 8, 14 bonds promoted by the 8 $\beta$ -hydroxy function as illustrated in Figure 1. The ms of compound **5**, the 24-methyl homologue of the hexol **1**, with ions at  $m/z$  225 (**a**, 40%), 239 (**b**-H<sub>2</sub>O, 100%) and 221 (**b**-2H<sub>2</sub>O, 68%), gives support to our suggestion. The pmr spectrum of **4** contained several features, two doublets of doublets at  $\delta$  4.01 ( $J=7.5$  and 3Hz) and 4.09 ( $J=10.5$  and 3Hz), this latter superimposed on the signal for 6 $\beta$ -H, and the A portion of an ABX system at  $\delta$  3.43 ( $J_{AB}=10.5$ Hz,  $J_{AX}=6$ Hz; the B portion, resonated under the MeOH signal) was already observed in the spectrum of the hexol **1** (7) and assigned to 16 $\alpha$ -H, 15 $\beta$ -H and 26-H, respectively. When we measured the spectrum of the derived tetraacetate **4a** the overlapping methine protons at  $\delta$  4.09 were shifted to lower field at  $\delta$  4.72 (15 $\beta$ -H) and 5.28. This latter signal had the characteristic shape (dt with  $J$  of 4 and 10.5 Hz) of the axial proton associated with the 6 $\alpha$ -acetoxyl group (9).

The spectrum of **4** also included one-proton double doublet at  $\delta$  2.50 ( $J=13$  and 3Hz), already observed in the hexol **1** and assigned to 7 $\beta$ -H. These data suggested the presence of a 6 $\alpha$ ,8 $\beta$ -dihydroxy moiety. Several features of the pmr spectra indicated a new hydroxyl group at 4 $\beta$ -position. The spectrum of the heptol **4** displayed an isolated signal at  $\delta$  4.28, which was quite narrow ( $W^{1/2}=6$ Hz) as would be expected for an equatorial proton and a methyl singlet at  $\delta$  1.22 (CH<sub>3</sub>-19) downfield shifted by 0.20 ppm relative to hexol **1** in agreement with the postulation of a 4 $\beta$ -OH in **4** (7, 10). The remaining three methyl signals,  $\delta$  0.94d, 0.96d, and 1.14s, were close to the values for CH<sub>3</sub>-21, CH<sub>3</sub>-27 and CH<sub>3</sub>-18, respectively, observed in the spectrum of the hexol **1** (7). Further, the pmr spectrum of the tetraacetate **4a** still contained overlapping hydroxyl and acetoxyl methine signals,  $\delta$  ca 4.72 (2H, 3 $\alpha$ -H, 15 $\beta$ -H) and 3.96 (3H, 4 $\alpha$ -H, 16 $\alpha$ -H and 26-H), but irradiation around 3.96 ppm in a double resonance experiment simplified the complexity of signal at  $\delta$  4.72 from which distinctly emerged a sharp

doublet ( $J=10.5$  Hz,  $15\beta\text{-H}$ ) and a signal clearly composed of four peaks, evidently a doublet of doublets with  $J=11.5$  and  $4$  Hz, as would be expected for the axial proton associated with the  $3\beta\text{-OH}$  still coupled with two other protons. On these bases, the position for the new hydroxyl group was narrowed down to C- $2\beta$  or C- $4\beta$ . The peak due to  $6\beta\text{-H}$  in **4** ( $\delta\sim 4.10$ , overlapping with  $15\beta\text{-H}$ ) and in **4a** ( $\delta$  5.28), was significantly downfield shifted by  $\sim 0.50$  and  $0.30$  ppm relative to the hexol **1** and its tetraacetate (3,6,15,26-tetraacetate), respectively (7). These data clearly show that the new hydroxyl group is located at position  $4\beta$ .

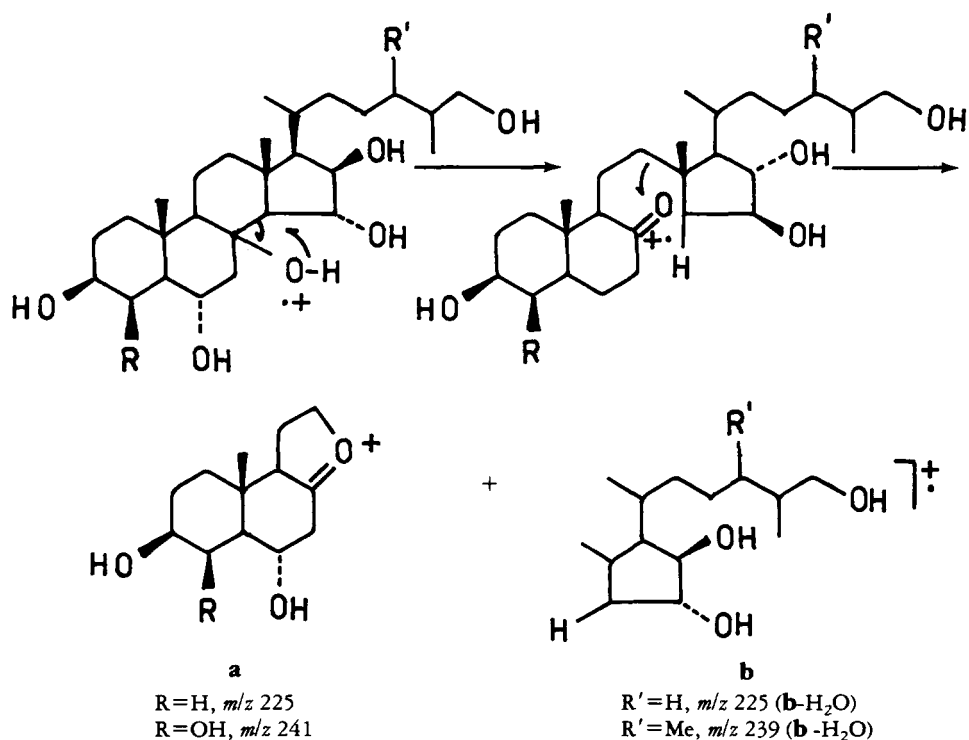


FIGURE 1. Fragments in the mass spectra of 8-hydroxysteroids.

During our investigation of biologically active marine sterols from Echinoderms, we have been working on the extractives of the Pacific starfish *Acanthaster planci* and have isolated in 10-mg amounts a polyhydroxylated sterol identical with **4**. Thus, it was possible to run a cmr spectrum and confirm that **4** was related to **1** by introduction of the seventh hydroxyl group at the  $4\beta$ -position. The cmr chemical shift assignments are shown in Table 1. The most significant features of the spectrum of **4**, which suggested the location of the new hydroxyl group at C- $4\beta$ , were the upfield shifts exhibited by C-2 (5.2 ppm) and C-6 (2.9 ppm), and the downfield shifts experienced by C-3

TABLE 1. Cmr Shifts in **4** (CD<sub>3</sub>OD)

C-1	39.9	C-10	38.0	C-19	17.0
C-2	26.3	C-11	18.9	C-20	30.6
C-3	73.6	C-12	43.2	C-21	18.3
C-4	69.2	C-13	45.4	C-22	37.2
C-5	57.2	C-14	64.6	C-23	24.8
C-6	64.7	C-15	80.8	C-24	35.0
C-7	50.5	C-16	83.1	C-25	37.0
C-8	76.0	C-17	60.7	C-26	68.6
C-9	58.4	C-18	16.9	C-27	17.2

(1.4 ppm), C-5 (3.5 ppm), and C-19 (2.8 ppm) relative to the hexol **1** (7). Similar shifts have been observed in 4 $\beta$ -hydroxysteroids (e.g., in 5 $\alpha$ -cholestane-3 $\beta$ ,4 $\beta$ -diol relative to 3 $\beta$ -cholestanol the  $\beta$  shifts at C-3 and C-5 are +0.8 and +4.0 ppm, respectively; the  $\gamma$ -shifts at C-2 and C-6 are -5.0 and -2.8 ppm, respectively; and the  $\delta$  shift at C-19 is +2.4 ppm (11, 12). Small shifts were also exhibited by C-9 and C-11, i.e., +1.0 and -0.5 ppm, respectively, relative to the hexol **1** (+1.0 and -0.7 ppm in 5 $\alpha$ -cholestane-3 $\beta$ ,4 $\beta$ -diol *vs* 3 $\beta$ -cholestanol; + denotes downfield shifts, and - denotes upfield shifts).

24 $\xi$ -Methyl-5 $\alpha$ -Cholestane-3 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-hexol (**5**) had mp 253-256°, [ $\alpha$ ]<sub>D</sub> = +27.8°. In the eims the highest molecular weight ion observed ( $m/z$  464) corresponded to loss of H<sub>2</sub>O from the molecular formula C<sub>28</sub>H<sub>50</sub>O<sub>6</sub>. In addition to intense peaks for stepwise H<sub>2</sub>O loss, the ms showed also the loss of the side-chain with peaks at  $m/z$  321, 303, 285, 267, identical to those that we had previously seen for the cholestane-hexol **1** (7), indicating an additional methyl group in the hydroxylated side-chain. The base peak at  $m/z$  239 (see Figure 1) confirmed the presence of an hydroxylated C<sub>9</sub>-side-chain. The pmr spectrum of **5** was very similar with that of the hexol **1** (7); the only difference being observed in the methyl region. In the spectrum of **5** there were two methyl doublets at  $\delta$  0.82 ( $J=6.5$ Hz) and 0.84 ( $J=6.5$ Hz) instead of the three protons of the methyl doublet observed at  $\delta$  0.91 in the spectrum of **1**, the remaining methyl signals, 0.95d, 1.05s, and 1.15s were close to the values for CH<sub>3</sub>-21, CH<sub>3</sub>-19, and CH<sub>3</sub>-18, respectively, observed in the spectrum of **1**.

Similarly, treatment of **5** with Ac<sub>2</sub>O and pyridine lead to a 3 $\beta$ ,6 $\alpha$ ,15 $\alpha$ ,26-tetraacetate, **5a**, whose pmr spectrum in the acetoxy, hydroxy-methine and methylene proton regions was superimposable with that of 5 $\alpha$ -cholestane-3 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-hexol 3,6,15,26 tetraacetate (7); in the methyl region the methyl doublets were seen at  $\delta$  0.80, 0.84, and 0.90. On this basis we suggest that the new sterol is the 24-methyl homologue of the hexol **1**. The location of the primary hydroxyl group at C-26 received support by double resonance experiments. Irradiation around 1.9 ppm in **5a** collapsed only the methyl doublet resonating at  $\delta$  0.90, and irradiation around 1.5 ppm collapsed the methyl doublet resonating at  $\delta$  0.84 leaving intact the doublet at  $\delta$  0.80. These experiments ruled out an isopropyl structure. The location of the "extra" methyl group at C-24 is mainly based on biogenetic grounds.

Upon catalytic hydrogenation, the compound **6**, whose  $\Delta^{22}$ -24-methyl structure was supported by pmr double resonance experiments, yielded a saturated derivative showing in the pmr spectrum three methyl doublets with chemical shifts  $\delta$  0.82, 0.85, and 0.95 virtually the same as in **5**.

(22E) 24-Methyl-5 $\alpha$ -Cholest-22-ene-3 $\beta$ ,4 $\beta$ ,6 $\alpha$ ,8,15 $\alpha$ ,16 $\beta$ ,26-heptol (**6**) had mp 185-188°, [ $\alpha$ ]<sub>D</sub> = +20.0°. In the eims the highest molecular weight ion observed ( $m/z$  478) corresponded to loss of H<sub>2</sub>O from the molecular formula C<sub>28</sub>H<sub>48</sub>O<sub>7</sub>. In **6**, the chemical shifts and coupling constants for the protons at position 3,4,6,15, and 19 were virtually the same as in **4** (Experimental). The C-16 proton was shifted upfield ( $\delta$  3.93 *vs.* 4.01) and the terminal C-26 protons were shifted downfield ( $\delta$  3.41, dd,  $J=10.5$  and 6.0Hz and  $\delta$  3.55, dd,  $J=10.5$  and 6.5 Hz). The singlet due to C-18 protons was also shifted downfield ( $\delta$  1.17 *vs.* 1.14). The pmr spectrum also showed the presence of a  $\Delta^{22}$  double bond, which is responsible for the above shifts, and a 24-methyl substituent in the side-chain. In the olefinic region a multiplet around 5.45 ppm and in the methyl region, doublets at  $\delta$  1.045 ( $J=6.5$ Hz), 0.99 ( $J=6.5$ Hz) and 0.91 ( $J=7$ Hz) are in agreement with a 26-hydroxy, 24-methyl- $\Delta^{22}$ -side-chain. The protons at C-20 and C-24 in an allylic position have also been shifted considerably downfield and are found as isolated multiplets at  $\delta$  2.55 and 2.10, respectively. Irradiation of the multiplet at  $\delta$  2.55 (20-H) in a double resonance experiment did indeed col-

lapse the doublet at  $\delta$  1.045 ppm to a singlet, and also simplified the complexity of the signal for the  $\Delta^{22}$  protons. Irradiation at 2.11 ppm (H-24) again simplified the  $\Delta^{22}$  pattern and collapsed the doublet at 0.99 ppm to a singlet.

The stereochemistry at the  $\Delta^{22}$  double bond is suggested to be *trans*. The C-21 methyl group signal is displayed in "normal"  $3\beta$ -hydroxy sterols at 1.00-1.01 ppm when it occurs together with a  $\Delta^{22}$ -*trans* double bond whereas a *cis* double bond shifts it to 0.94-0.95 ppm (13). Our value of 1.045 ppm seems indicative that the double bond is *trans* in **6**.

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#### LITERATURE CITED

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